

1. Calculate the energy of the photon, given the frequency ($\nu = 2 \times 10^9$):

$$E = h \nu = 4.136 \times 10^{-15} \bullet 2 \times 10^9 = 8.27 \times 10^{-6} \text{ eV}$$

$$= 6.626 \times 10^{-34} \bullet 2 \times 10^9 = 1.325 \times 10^{-24} \text{ Joules}$$

2. What is the frequency of a 1 mm wave?

$$\lambda f = c; \quad f = \frac{3 \times 10^8}{1 \times 10^{-3}} = 3 \times 10^{11} \text{ Hz}$$

3. Black - body radiation problem:

$$T = 20 + 273 = 293 \text{ K}$$

$$\text{Power / area} = \sigma T^4 = 5.67 \times 10^{-8} \bullet 293^4 = 417.9 \text{ Watts / m}^2$$

$$\text{Power} = 4\pi r^2 \bullet 417.9 = 5.25 \text{ kiloWatts}$$

4. At what wavelength does the radiation peak?

$$\text{For } T = 20 \text{ C} = 293 \text{ K}$$

$$\text{For } T = 1000 \text{ C} = 1273 \text{ K}$$

$$\lambda_{\max} = \frac{a}{T} = \frac{2.898 \times 10^{-3}}{293}$$

$$= 9.89 \times 10^{-6} \text{ m}$$

$$\lambda_{\max} = \frac{a}{T} = \frac{2.898 \times 10^{-3}}{1273}$$

$$= 2.277 \times 10^{-6} \text{ m}$$

5. Calculate the equilibrium temperature of a satellite in free space, at a distance from the sun of 1 astronomical unit (1 AU = 93 million miles = 1.5×10^8 km, Solar Radius $R_{\odot} = 6.96 \times 10^5$ km). Take the emissivity, ϵ , to be one for the sun (a good assumption), and the satellite (less so).

a) Calculate the radiated power: $P = \sigma T_{\text{sun}}^4 \bullet 4\pi R_{\text{sun}}^2$

$$P = 5.67 \times 10^{-8} \bullet 5800^4 \bullet 4\pi (6.96 \times 10^8)^2 = 3.91 \times 10^{26}$$

- b) Calculate the amount of power which reaches earth - it drops off as Gauss' law might suggest:

$$P = \sigma T_{\text{sun}}^4 \bullet 4\pi R_{\text{sun}}^2 \bullet \frac{1}{4\pi R_{\text{Earth Orbit}}^2}$$

$$P = 5.67 \times 10^{-8} \bullet 5800^4 \bullet \frac{4\pi (6.96 \times 10^8)^2}{4\pi (1.5 \times 10^{11})^2} = 1381 \frac{\text{Watts}}{\text{m}^2}$$

You can check that this gives the known answer: $1370 \pm 4 \text{ W/m}^2$

c) The radiation incident on the satellite must be re-radiated:

$$P_{\text{incident}} = \sigma T_{\text{sun}}^4 \cdot 4\pi R_{\text{sun}}^2 \cdot \frac{1}{4\pi R_{\text{Earth Orbit}}^2} \cdot \pi R_{\text{satellite}}^2 = \sigma T_{\text{satellite}}^4 \cdot 4\pi R_{\text{satellite}}^2$$

noting that the radiation is incident on the projected area of the sphere, which is the circular disk cross-section. Solve the last equation for the satellite temperature.

$$T_{\text{sun}}^4 \cdot R_{\text{sun}}^2 \cdot \frac{1}{4R_{\text{Earth Orbit}}^2} = T_{\text{satellite}}^4$$

$$T_{\text{satellite}} = T_{\text{Sun}} \cdot \left[\frac{R_{\text{sun}}^2}{4R_{\text{Earth Orbit}}^2} \right]^{\frac{1}{4}} = T_{\text{Sun}} \cdot \left[\frac{R_{\text{Sun}}}{2R_{\text{Earth Orbit}}} \right]^{\frac{1}{2}}$$

$$T_{\text{satellite}} = 5800 \cdot \left[\frac{6.96 \times 10^8}{2 \cdot 1.5 \times 10^{11}} \right]^{\frac{1}{2}} = 5800 \cdot 0.048 = 279 \text{ K}$$

6. For a Bohr atom, find the energy for the $n=3$ to $n=2$ transition.

$$\Delta E_{\text{electron}} = E_f - E_i = -13.58 \left(\frac{1}{n_f^2} - \frac{1}{n_i^2} \right) = -13.58 \left(\frac{1}{4} - \frac{1}{9} \right) = -1.886 \text{ eV}$$

the photon energy is then the negative of this, or $+1.886 \text{ eV}$

$$E_{\text{photon}} = h f = 4.136 \times 10^{-15} \cdot f = 1.886 \text{ eV}; \quad f = 4.56 \times 10^{14}$$

$$\lambda f = c; \quad \lambda = \frac{3 \times 10^8}{4.56 \times 10^{14}} = 6.58 \times 10^{-7} \text{ meters}$$

This is termed the H-alpha (H_{α}) transition

7. Start by converting to Joules:

$$kT = 1.38 \times 10^{-23} \cdot 1000 = 1.38 \times 10^{-20} \text{ J}$$

The average kinetic energy of one atom is then given by:

$$\frac{1}{2} m v^2 = \frac{3}{2} kT = 1.5 \cdot 1.38 \times 10^{-20} \text{ J} = 2.07 \times 10^{-20}$$

Note that the factor of 1.5 is really a detail....We take the atom to be hydrogen

$$v = \sqrt{\frac{2E}{m}} = \sqrt{\frac{2.07 \times 10^{-20}}{1.67 \times 10^{-27}}} = 5.0 \times 10^3 \text{ m/s}$$

what would the velocity be for oxygen? (mass = $16 \cdot 1.67 \times 10^{-27}$)

$$8. n(K) = \frac{2N \sqrt{K} e^{-K/kT}}{\sqrt{\pi}(kT)^{3/2}} = \frac{2N}{\sqrt{\pi}} \frac{1}{(kT)} \sqrt{\frac{K}{kT}} e^{-K/kT};$$

having rewritten the equation slightly, it should now be obvious that the dimensions are number per energy - ultimately to get number you would integrate over all energies.

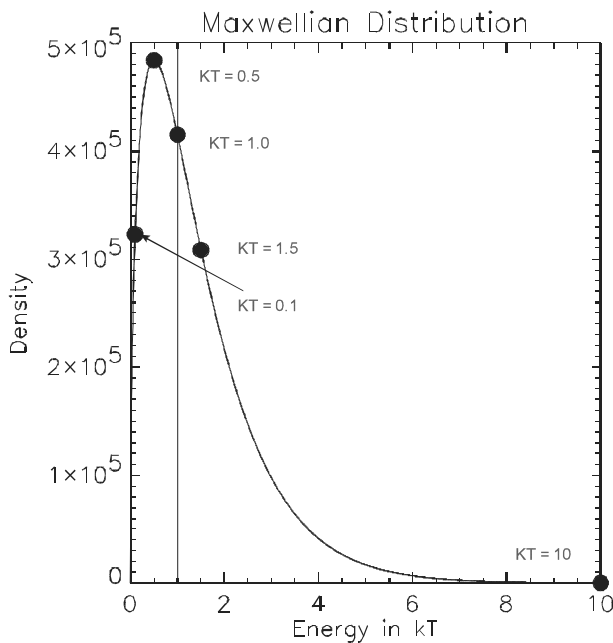
Taking $N = 10^6$, $kT = 1$.

$$\begin{aligned} n(0.1 \text{ kT}) &= \frac{2N}{\sqrt{\pi}} \frac{1}{(kT)} \sqrt{\frac{K}{kT}} e^{-K/kT} = \frac{2 \cdot 10^6}{\sqrt{\pi}} \frac{1}{(1)} \sqrt{\frac{K}{kT}} e^{-K/kT} \\ &= \frac{2 \cdot 10^6}{\sqrt{\pi}} \frac{1}{(1)} \sqrt{0.1} e^{-0.1} = 1.13 \times 10^6 \cdot \sqrt{0.1} e^{-0.1} = 1.13 \times 10^6 \cdot 0.286 = 3.23 \times 10^5 \end{aligned}$$

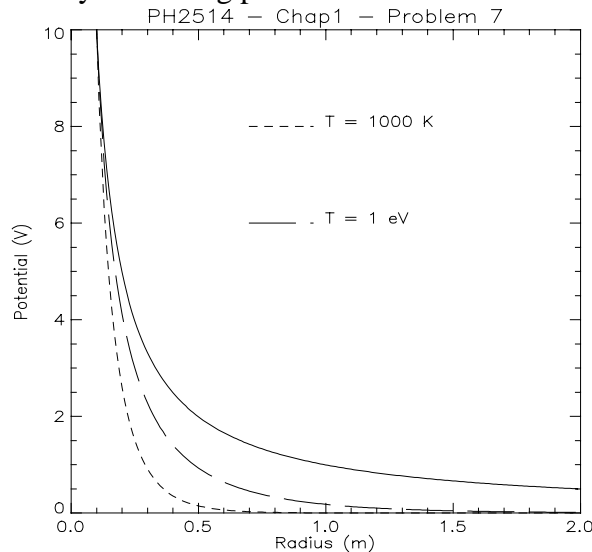
$$n(0.5 \text{ kT}) = \frac{2 \cdot 10^6}{\sqrt{\pi}} \frac{1}{(1)} \sqrt{0.5} e^{-0.5} = 1.13 \times 10^6 \cdot 0.429 = 4.84 \times 10^5$$

$$n(1.5 \text{ kT}) = \frac{2 \cdot 10^6}{\sqrt{\pi}} \frac{1}{(1)} \sqrt{1.5} e^{-1.5} = 1.13 \times 10^6 \cdot 0.273 = 3.08 \times 10^5$$

$$n(10 \text{ kT}) = \frac{2 \cdot 10^6}{\sqrt{\pi}} \frac{1}{(1)} \sqrt{10} e^{-10} = 1.13 \times 10^6 \cdot 1.44 \times 10^{-4} = 1.62 \times 10^2$$



9. Debye shielding problem...



The general formula is:

$$\lambda_D \text{ (meters)} = \left(\frac{\epsilon_0 kT}{e^2 n_e} \right)^{1/2}$$

but it is easier to use the numerical forms:

(A) $n = 1.0 \times 10^{11}$, $T = 1000$ K;

$$\lambda_D = 69 \left(\frac{T}{n_e} \right)^{1/2} = 69 \left(\frac{1000}{1.0 \times 10^{11}} \right)^{1/2} \\ = 69 \cdot 10^{-4} \text{ m} = 0.69 \text{ cm};$$

and,

(B) $n = 1.0 \times 10^6$, $T = 500$ eV

$$\lambda_D = 7430 \left(\frac{T}{n} \right)^{1/2} = 7430 \left(\frac{500}{1.0 \times 10^6} \right)^{1/2} = 166 \text{ m}$$

now just plot up $\Phi = \frac{1}{4\pi\epsilon_0} \frac{q}{r} e^{-(r-r_0)/\lambda_D}$

Note that the solid line in the plot is for no Debye sheath (just a coulomb potential)

11. $B = 100$ nano-Tesla,

$$B = 100 \times 10^{-9} \text{ T}, f = \frac{qB}{2\pi m} = \frac{1.6 \times 10^{-19} \cdot 100 \times 10^{-9}}{2\pi \cdot 1.67 \times 10^{-27}} = 1.525 \text{ Hz}$$

12. Use $B = 3.1 \times 10^{-5}$ Tesla is the canonical value

$$f = \frac{qB}{2\pi m} = \frac{1.6 \times 10^{-19} \cdot 3.1 \times 10^{-5}}{2\pi \cdot 9.1 \times 10^{-31}} = 0.876 \times 10^6 \text{ Hz} \approx 1 \text{ MHz}$$

13. Drift velocity (E cross B). Note that the drift velocity is independent of energy - the value of 5 eV is extraneous information.

$$E = \frac{\mathbf{E} \times \mathbf{B}}{B^2} = \frac{E}{B} = \frac{10^{-3}}{200 \times 10^{-9}} = 5000 \frac{\text{m}}{\text{s}} = 5 \text{ km/s}$$

14 Plasma pressure...

$$p = \sum n_i kT_i = 2(2.0 \times 10^{10} \cdot 0.5) = 2.0 \times 10^{10} \text{ eV/m}^3 \\ = 1.6 \times 10^{-19} \cdot 2.0 \times 10^{10} = 3.2 \times 10^{-9} \text{ J/m}^3$$

(note that the extra "2" above comes from having **both** ions and electrons.

The magnetic energy density (pressure) is given by:

$$p = \frac{B^2}{2\mu_0} = \frac{(3 \times 10^{-6})^2}{2 \cdot 4\pi \times 10^{-7}} = 3.58 \times 10^{-6} \text{ J/m}^3$$

The ratio is $\beta = \frac{\sum n_i k T_i}{B^2 / 2\mu_0} = \frac{3.2 \times 10^{-9}}{3.58 \times 10^{-6}} = 8.94 \times 10^{-4}$, so the magnetic field

dominates